STEREOSPECIFIC EPOXIDATION OF ANTHRACYCLINONES A POUTE TO 8(R)-MFTHOXYDAUNOMYCINONE

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<u>Abstract</u> - Sterecoelective epoxidation of <u>1</u>, the 7,8-unsaturated derivative of daunomycinone, followed by <u>trans</u> opening of the epoxide <u>2</u>, afforded 8(R)-methoxydaunomycinone <u>6a</u>, its configuration at C-8 was determined by chemical correlations and PMR studies

As part of our continuing investigation into the synthesis of new derivatives of the antitumouranthracyclines daunorubicin and doxorubicin modified at ring A,¹ we wish to report the synthesis and stereochemistry of 8(R)-methoxydaunomycinone <u>6a</u> Among the anthracyclinones, the only reported compound substituted at C-8 is the aglycone of steffimycin A and B, bearing in such position a methoxyl group,² whose configuration, however, was not determined

Unsaturated daunomychnone derivative $\underline{1}^{3,4}$ is a useful chiral intermediate for the introduction of a methoxyl group at C-8 of daunomychnole (see scheme) The epoxy derivative $\underline{2a}$ mp 166-170°C (dec) obtained almost quantitatively froi $\underline{1}$ with <u>m</u>-chloroperbenzoic acid (CHCl₃, R T) was converted (KI, PhCH₂Br DMF-Me₂CO, 60°C for 3 hrs) in 70% yield after chromatography into dibenzyl ether $\underline{2b}$ mp 38-89°C 5

opening of the oxirane ring by treatment of <u>2b</u> with methanol and a catalytic amount of <u>p</u>-toluenesulfonic acid gave only two products (9 1 ratio) of which <u>3a</u> was the main one, mp 102-104°C.⁶ The minor product <u>4a</u>, mp 117-120°C, differed from <u>3a</u> only in the stereochemistry at C-13 In fact both products gave only <u>5a</u> upon mild acidic treatment (aq 80% TFA, 0°C, 5 min). Compounds <u>3b</u>, mp 105-107°C, and <u>4b</u>, mp 205-207°C, obtained from <u>2a</u> (MeOH, <u>p</u>-toluenesulfonic acid, 5 hrs at R T, 80% yield after chromat.raphy) in 6 1 ratio, as well as the corresponding accetates <u>3c</u> (Ac₂0, pyridine, R T, 16 hrs), mp 225°C, and <u>4c</u>, mp 260°C, were useful in the determination of the S configuration at C-13 in compounds <u>3a,c</u>. In fact the CH₃C(<u>OCH₃</u>) signal shifted from 3.43 to 3.26 **s** in converting <u>3b</u> into <u>3c</u>, whereas a similar change was not observed for the pair of compounds <u>4b</u> (3.34**s**) and <u>4c</u> (3.35**s**) This interpretation was corroborated by the chemical shift difference between H-10eq and H-10ax, which was larger in <u>4a c</u> ($\Delta \delta = 0.4$) than in 3a, c ($\Delta \delta = 0.1$)

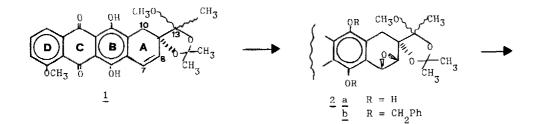
Methylation of the hydroxyl group of <u>3a</u>, (MeI and NaH, THF, 55°C for 6 hrs), gave <u>3d</u>, mp 105-107°C, PMR <u>1 a</u>. § 3.37 (s, CH_3^{0-8}), 4 00 (d, J = 2 Hz, H-8) Finally, treatment of <u>3d</u> with TFA containing a tiny amount of water at R T for 24 hrs, afforded <u>6a</u>, <u>7a</u>, and <u>9</u>, separated by silicagel column chromatography. Compounds <u>6a</u> and <u>7a</u> differed in the stereochemistry at C-7 indeed by catalytic benzylic hydrogenolysis (5% Pd/BaSO₄, 1 atm, R T) both gave <u>8</u> as unique product. 2253 <u>c</u>

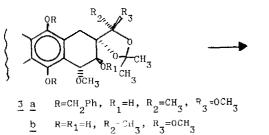
d

<u>4 a</u>

b

<u>c</u>



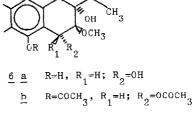


 $R=R_1=COCH_3$, $R_2=CH_3$, $R_3=OCH_3$

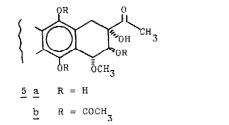
 $R=CH_2Ph$, $R_1=R_2=CH_3$, $R_3=OCH_3$

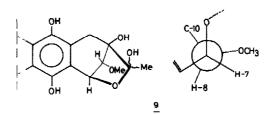
 $R=R_1=H$, $R_2=OCH_3$, $R_3=CH_3$ $R=R_1=COCH_3$, $R_2=OCH_3$, $R_3=CH_3$

 $R-CH_2Ph$, $R_1=H$, $R_2=OCH_3$, $R_3=CH_3$



$$\begin{array}{ccc} \underline{7} & \underline{a} & R=H, R_1=0H, R_2=H \\ \underline{b} & R=COCH_3, R_1=OCOCH_3, R_2=H \\ \underline{8} & R=R_1=R_2=H \end{array}$$





Compound <u>6a</u>, mp 238°C, $/ \alpha / D + 147°$ (c 0 107, CHCl₃), PMR <u>1 a</u>. δ 3 04 (d, J = 17 Hz, H-10ax), 3 28 (d, J = 17 Hz, H-10eq), 3 51 (s, CH₃0-8), 3 71 (d, J = 3 Hz, H-8), 5 15 (d, J = 3 Hz, H-7), by acidic treatmont with 2,2-dimethoxypropane afforded the 7,9-isopropylidene derivative, mp 206-207°C, PMR <u>1.a</u> 50 96, 1.30 (two s, C(CH₃)₂), 2.27 (s, COCH₃), 4.14 (d, J = 4 Hz, H-8), 5.30 (d, $J = 4 H^{\prime}$, H-7), thus showing the cis diaxal relationship of 0H-7 and 0H-9. The value of J did not allow to distinguish between H-7eq, H-8eq and H-7eq, H-8ax. However the formation of small amounts of 9 from 7a, via internal acetalyzation between 0H-7 and the C-13 carbonyl group, allowed to establish the orientation of H-8 in 7a In fact, in 9, PMR 1.a. 6 1 31 (bs, $C(OH)CH_{,}$, 5.60 (d, J = 2.5 Hz, H-7), the formation of the five membered ring induces a change of conformation of ring A and a deformation of the dihedral angle between H-7 and H-8 as compared with that in the unstrained isopropylidene derivative of 6a, thus causing the observed change of J from 4.0 to 2 5 Hz, which can be justified only by assigning H-8 to the α axial position H-7.H-8 (see Newman projection of 9). This is in agreement with a H-Tax, H-8eq relationship in the parent compound <u>7a</u>. The above interpretation is supported by the values of $J_{H-7,H-8}$ in the acetyl derivatives $\frac{6b}{H}$, mp 265°C, and $\frac{7b}{H}$, mp 205°C (J_H-7eq, H-2eq - 1 6 Hz and J_H-7ax, H-8eq = 4.5 Hz, respectively)

Since the configuration of the chiral centre at C-8 is directly related to the stereochemistry of epoxide 2a, this is β with respect to the plane of ring A, and since 2 is the unique product, the epoxidation reaction of 1 is stereoselective. Moreover, the similarity of CD curves of 5a and 6a with those of daunomycinone^{7,8} and other anthracyclinones^{8,9} allows to establish the S configuration at C-7 in 5a, thus revealing that in our conditions the oxirane ring of 2b undergoes exclusively a trans opening reaction.

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References and Notes

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- 3) F Arcamone, and G Franceschi, unpublished results Compound <u>1</u> was obtained as a mixture of epimers at C-13 in 28% yield after chromatography by treatment of daunomycinone with 2,2-dimethoxypropa e and <u>p</u>-toluenesulfonic acid in anhydrous $CHCl_3$; mp 290-295°C (dec); \bigwedge_{max} (CHCl_3) 385, 507, and 542 nm, PMR <u>1 a</u> f 1 25-1 52 (four s, $C(CH_3)_2$, $C(OCH_3)CH_3$), 3 27, 3 36 (two s, $C(CH_3)OCH_3$), 6 26, 6 32 (two d, J = 10 Hz, H-8), 7 10, 7.16 (two d, J = 10 Hz, H-7), 12 97, 13 00, 13 37, and 13 50 (four s, OH-6 and OH-11)
- 4) All compounds gave mass, IR, and PMR (CDC1₃) spectra consistent with the assigned structures Yields are unoptimized and mp uncorrected
- 5) PMR of $\underline{2b}$ <u>1</u> <u>a</u>. $\mathbf{\delta}$ 1 38, 1 53 (two s, $C(CH_3)_2$, $C(OCH_3)\underline{CH}_3$, 3 32 (s, $C(CH_3)\underline{OCH}_3$), 3 98 (s, $CH_3\mathbf{0}-4$, H-8), 4.37 (d J = 4 5 Hz, H-7). However, in the PMR spectrum of $\underline{2a}$, either before benzylation or when reobtained upon hydrogerolysis (5% Pd/BaSO₄, 1 atm, R Γ) of $\underline{2b}$, the signals of $C(CH_3)\underline{OCH}_3$ (3 32 and 3.36 $\mathbf{\delta}$) and of H-7 (4.54 and 4 59 $\mathbf{\delta}$) were splitted, indicating the presence of two C-13 epimers in 1 1 ratio
- 6) PMR of <u>3a</u> <u>1.a.</u> 6 1.26, 1.40, 1 46 (three s, $C(\underline{CH}_3)_2$, $C'OCH_3)\underline{CH}_3$), 2 82 (d, J = 16 Hz, H-10ax), 3 24 (d, J = 16 Hz, H-10eq), 3 32 (s, $C(CH_3)O\underline{CH}_3$), 3 37 (s, CH_3O-7), 4.52 (d, J = 3 5 Hz, H-8), 4 80 (d, J = 3 5 Hz, H-7)
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